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$\Delta S^\circ = [197.6 + 130.6] - [5.74 + 188.7] = 133.8 \text{ J/K}$   
 $\Delta G^\circ = [-137.2 + 0.0] - [0.0 + -228.6] = 91.4 \text{ kJ}$   
 $\Delta G^\circ = 131.3 - (298)(0.1338) = 91.4 \text{ kJ}$ . Both methods give the same  $\Delta G^\circ$ , as indeed they should.

### CHM 112 Thermodynamics Practice Problems Answers

Activity 1.3.3 Thermodynamics Answer Key Subject: POE - Unit 1 - Lesson 1.3 - Energy Applications Author: POE Revision Team Last modified by: Jason Rausch Created Date: 5/18/2015 3:08:00 PM Company: Project Lead The Way, Inc. Other titles: Activity 1.3.3 Thermodynamics Answer Key

### Activity 1.3.3 Thermodynamics Answer Key

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